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SYNTHESIS OF 2-CHLORO- AND 2,5-DICHLOROBENZOPHENONES

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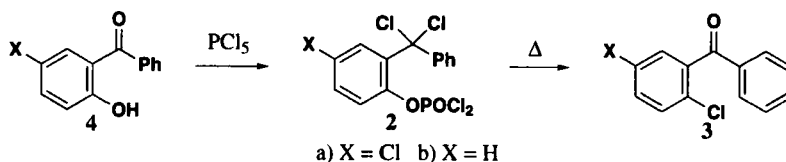
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SYNTHESIS OF 2-CHLORO- AND 2,5-DICHLOROBENZOPHENONES

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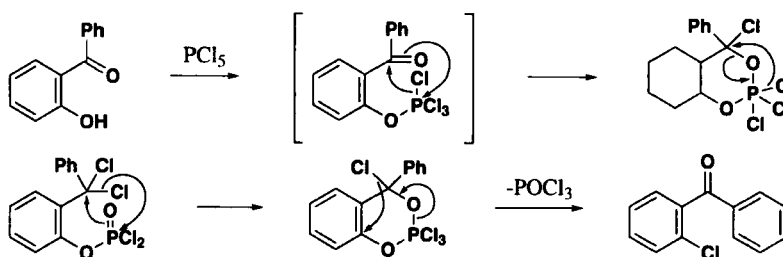
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2-Chloro- and 2,5-dichlorobenzophenones (**3b** and **3a**) were required in connection with another project. The yields of the reported syntheses of 2,5-dichlorobenzophenone (**3a**) from the Friedel-Crafts reaction of *p*-dichlorobenzene with benzoyl chloride are either not reported¹ or low (20%² and 8%³) and these preparations involve long reaction times and vigorous conditions. In our own preliminary work, only a 17% yield of **3a** was obtained *via* the Friedel-Crafts reaction after 49 hrs. The low yields probably result from the ring-deactivating effect of the two chlorine substituents. The synthesis reported here produces **3a** from **1a** in two steps, each proceeding in high yields. The compound formed in the first step (**2a**) need not be isolated and may be heated directly after formation; it should be noted that both steps are new reactions (*Scheme 1*).



Scheme 1

Possible rationalizations for the reaction of **1** with PCl_5 and for the elimination of phosphorus oxychloride from **2** on heating may be depicted as shown in *Scheme 2*.⁴ An esr study of the pyrolysis reaction indicated that no detectable radicals were generated in the reaction.⁵



Scheme 2

Although *o*-chlorobenzophenone was obtained in good yield by the Friedel-Crafts reaction of *o*-chlorobenzoyl chloride with benzene, it could also be generated in good yield (80%) by use of the present procedure. It is evident that a wide number of compounds, otherwise difficult

obtainable by other routes, could be accessed by the above sequence of reactions from the appropriate *o*-hydroxybenzophenones.

EXPERIMENTAL SECTION

Mps were obtained on a Thomas-Hoover Unimelt apparatus with T-H mps standards and are uncorrected. *o*-Hydroxybenzophenone, mp. 37-78°C, was prepared by a Friedel-Crafts reaction of *o*-anisoyl chloride with benzene by the procedures of Ullman and coworkers.⁶ Other compounds are commercially available. Carbon and hydrogen analyses were performed by Clark Microanalytical Laboratories, Urbana, IL; chlorine was determined by the silver thiocyanate-ferric alum method.

Synthesis of 2,5-Dichlorobenzophenone (3a).- Treatment of 5-chloro-2-hydroxybenzophenone with phosphorus pentachloride gave 5-chloro-2-(α,α -dichlorobenzyl)phenyl-phosphochloridate (**2a**) in 99% yield as previously described.⁷ This compound (3.92 g, 9.69 mmol) was heated in a distilling flask (protected from moisture) and POCl₃ (1.25 g, 95%) distilled at about 109-120°C. Heating the residue under reduced pressure (ca. 1 mm) gave the product **3a** (0.93 g, 40%), bp. 240-260°C, as a colorless liquid which solidified upon cooling. Recrystallization from ethanol gave 0.93 g (40%) of 2,5-dichlorobenzophenone, mp. 88.5-89.5°C, *lit.*¹⁻³ 88°C, 85-86°C, 87°C. An IR spectrum of the residue showed that it consisted mainly (> 92%) of the expected product. Repetition of the reaction of **2a** with PCl₅ on a larger scale and without isolation of **2a** gave an 86% yield of **3a** and an 84% yield of POCl₃.⁸

Synthesis of *o*-Chlorobenzophenone (3b). Preparation of *o*-(α,α -Dichlorobenzyl)phenylphosphochloridate (2b).- In a reaction vessel protected from moisture and arranged for trapping of HCl, a solution of *o*-hydroxybenzophenone (6.33 g, 31.9 mmol) in 60 mL of benzene was added over a period of 1 hr to PCl₅ (6.64 g, 31.9 mmol) covered with benzene (25 mL) at rt. After the addition was complete and further stirring for 30 min, the solvent was removed *in vacuo* to give a colorless liquid (11.7 g, 99%) which solidified on cooling, mp. 64-66°C. Recrystallization of a small sample from cyclohexane afforded colorless crystals, mp. 66-67°C.

IR (CCl₄): 1486, 1302, 1196, 940, 919 cm⁻¹. MW (cryoscopic): Calcd 370. Found 373.

Anal. Calcd for C₁₃H₉Cl₄O₂P: C, 42.19; H, 2.45; Cl, 38.33. Found: C, 42.08; H, 2.62; Cl, 38.21

The pyrolysis of **2b** was carried in an oil bath as described above, at 220°C to give an 80% yield of *o*-chlorobenzophenone (**3b**), mp. 46-47°C, *lit.*⁹ 46°C (and 77% of POCl₃). An authentic sample of *o*-chlorobenzophenone, mp. 46-47°C, was prepared in 90% yield by the Friedel-Crafts reaction of *o*-chlorobenzoyl chloride with benzene.

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REFERENCES

1. T. de Crauw, *Rec. Trav. Chim. Pays-Bas*, **50**, 753 (1931).
2. J. Ganzmuller, *J. prakt. Chem.*, **138**, 311 (1933).
3. G. Pfister-Guillouzo, M. Grimaud and J. Deschamps, *Bull. Soc. Chim. Fr.*, 1212 (1969).
4. Several attempts to chlorinate **2b** to its ring-chlorinated derivative (**2a**, X = Cl) failed, most likely because of the strong deactivation of the ring by the -CCl₂Ph groups.
5. C. Graebe and F. Ullman, *Ber.*, **29**, 824 (1896); F. Ullman and I. Goldberg, *Ber.*, **35**, 2811 (1902).
6. This rationalization is based on kinetic studies and the fact that there was no exchange with added phosphorus-32 oxychloride.
7. A. G. Pinkus and L. Y. C. Meng, *J. Org. Chem.*, **31**, 1038 (1966).
8. One referee suggested an alternative synthesis by conversion of commercially available 2,5-dichlorobenzoic acid to its acid chloride followed by a Friedel-Crafts reaction with benzene.
9. J. F. Bunnett and B. F. Hrutford, *J. Org. Chem.*, **27**, 4152 (1962)
